Warm Up -Atmosphere Unit

Turn your measuring homework in -place in your class' bin

-the state of the earth's 1.) What is weather?

atmosphere at a particular place and

2.) What do you know about the earth's atmosphere?

-the atmosphere is a thin layer of gases that surround the planet

3.) What do you want to learn about the atmosphere?

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Key Vocabulary

Air Pressure -the force exerted by air, whether compressed or

unconfined, on any surface in contact with it

-the height of anything above a reference level, Altitude

especially above sea level on earth

-the gaseous mass or envelope surrounding the earth, **Atmosphere**

or any celestial body, and retained by the earth's

gravitational field

Composition -the combining of distinct parts or elements to

form a whole

-the measure of how much mass per volume: m/v Density

Key Vocabulary

-a form of oxygen (O3) found in the Ozone

troposphere; can be good or bad

-layer of O3 in the atmosphere that absorbs Ozone Layer

UV rays and prevents it from reaching

Earth's surface

-Scientific properites are energy, matter, **Properties**

space, and time

-a measure of the warmth of coldness of an Temperature

object of substance with reference to some

standard value

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The Air Around You

The Earth's atmosphere is the covering of gases that surrounds Earth. The atmosphere (also called the air) is a very thin covering.

Weather is the condition of the Earth's atmosphere at a particular place and time. The troposphere is the layer of the Earth where weather OCCUrs.

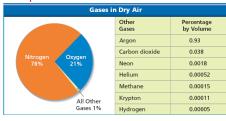
The History of our Atmosphere

Composition of the Atmosphere

Earth's atmosphere is made up of nitrogen, oxygen, carbon dioxide, water vapor, and many other gases, as well as particles of liquids and solids. Nitrogen is the most

Oxygen is the second most abundant gas; it makes up about 21% of the atmosphere.

Nitrogen is the most abundant gas in the atmosphere; it makes up 78% of the atmosphere.

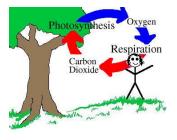


<-- Copy the pie chart and list the other gases present in our air. You don't have to include the percentages.

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Carbon dioxide is essential to life. Plants must have carbon dioxide to produce food. When the cells of animals break down food to produce energy, they give off carbon dioxide as a waste product. That carbon dioxide is then transported to our lungs where we get rid of it by exhaling.



Warm Up

- 1.) What instrument do we use to measure air pressure?
- 2.) What do you think happens to the air pressure and temperature as altitude increases?
- 3.) What do you think the air is like on top of Mt. Everest?

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Properties of Air

Air consists of atoms and molecules, which have mass. Because <u>air has mass</u>, it also has other properties, including density and pressure.

Pressure: The force pushing on an area or surface. The weight of the atmosphere exerts a force on surfaces. Air pressure is the result of the weight of a column of air pushing down on an area.

Barometer (buh rahm uh tur): an instrument that is used to measure air pressure. Two common kinds of barometers are mercury barometers and aneroid barometers.

A denser substance has more mass per unit volume than a less dense one. So denser air exerts more pressure than less dense air.



Let me quickly explain...

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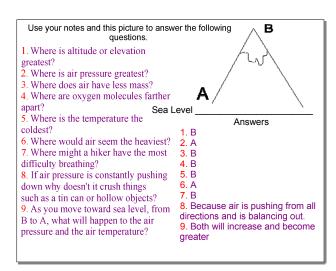
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Think of a column air that stretches above you that is approximately 300 miles high from sea level. As you move to higher ground, the column above you becomes less. Click paragraph above for demo... Click picture for 7 minute video...

3 Main Points!!!

- As altitude increases, air pressure decreases
- As altitude increases, temperature decreases
- As air pressure decreases, so does density

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As altitude increases, temperature decreases. At higher altitudes the air is thinner and colder than at lower altitudes. The air drops 3° for every 1,000 feet of altitude above sea level. (This rule is not accurate for altitudes above 40,000 ft)

Calculate the temperature at each altitude given.

Temperature	Altitude	Temperature @ altitude
32°	1,000 feet	29°
47°	3,300 feet	37°
66°	2,600 feet	58°
76°	39,999 feet	-44°
21°	16,900 feet	-29°

(answers are rounded)

Sep 10-4:35 PM

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<u>Ultraviolet Radiation background</u> information:

Sunlight is the greatest source of UV radiation. Man-

made ultraviolet sources include several types of UV

lamps. UV radiation is commonly used in industrial

processes and in medical and dental practices for a

fluorescent effects, welding, phototherapy and sun

variety of purposes, such as killing bacteria, creating

Warm Up!

1.) When referring to UV radiation, what does the "UV" stand for?

<u>u</u>ltra<u>v</u>iolet

2.) Where does UV radiation come from?

the sun

3.) What could happen if you

sunburn, skin cancer, eye damage

are exposed to too much UV radiation?

Sunscreen, hats, sunglasses,

4.) What might protect us from UV radiation?

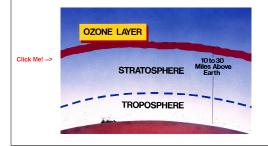
ozone

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tanning.

The sun radiates energy in a wide range of wavelengths, most of which are invisible to human eyes. The shorter the wavelength, the more energetic the radiation, and the greater the potential for harm. Living things and the cells they are made of are protected from large amounts of UV radiation by a chemical called **ozone**. A layer of ozone in the upper atmosphere absorbs UV radiation and prevents most of it from reaching the Earth. Without the layer of ozone in the stratosphere to protect us from excessive amounts of UV-B radiation, life as we know it would not be the same. Ozone can be good or bad, depending on where it is found.

Most oxygen molecules have two oxygen atoms. Ozone is a form of oxygen that has three oxygen atoms in each molecule instead of the usual two.



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Good vs. Bad

Good ozone (stratospheric ozone)- shields us from the suns UV rays. It is found 10-30 miles above Earth's surface in the stratosphere.

Bad ozone (tropospheric ozone)- found in the lower atmosphere, near the ground, is formed by cars, power plants, chemical plants and other sources that react in the presence of sunlight. It is a greater concern in areas with higher temperatures and higher pollution.

Who is at the highest risk for damage from the ozone?

- Children who spend a lot of time playing outdoors in the summer
- Anyone who has asthma or other respiratory disorders.
- Adults of all ages who work vigorously outside.

Ozone is like sunburn for the lungs. The only difference is that the lungs have no nerves, so you can't feel the pain.

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Dear 16 Year Old Me

This is a video about skin cancer and how it can affect us. The video will also discuss ways we can work to prevent skin cancer and what signs to look out for.

This video can be emotional, especially if you've known someone affected by skin cancer. If at any moment you want/need to leave the room due to the video, please do.

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